Safety Data Sheet Potassium Silicate CP 40



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Section	1.	Chamical	Droduct	and	Company	/ Identification
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Product Name: Potassium Silicate CP 40

CAS No.: 1312-76-1

EC/EINECS: 215-199-1

UN No: -

EC Annex1 Index No: -

Chemical Name: Potassium Silicate

Synonyms : Water glass, Silicic acid, Sodium salt

	Section 2: Hazards Identification
GHS Classification:	Oral acute toxicity category 5
	Skin irritation category 3
	Eye irritation category 2B
GHS Label:	No label
GHS Signal Word:	Warning
GHS Precaution	H303 May be harmful if swallowed
Phrases:	H316 Causes mild skin irritation
	H320 Causes mild eye irritation
GHS Response	P312 Call a doctor if you feel unwell.
Phrases:	P332 + P313 If skin irritation occurs: Get medical advice/ attention.
	P305 + P351 + P338 If in eyes: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.
	P337 + P313 If eye irritation persists: Get medical advice/ attention.
Emergency Overview:	Clear, colorless, odorless, thick liquid. Causes mild eye, skin, and digestive tract irritation. Spray mist causes mild irritation to respiratory tract. Noncombustible. Spills are slippery. Reacts with acids, ammonium salts.
Chronic hazards:	No known chronic hazards. Not listed by NTP, IARC or OSHA as a carcinogen.
Physical hazards:	Dries to form glass film which can easily cut skin. Spilled material is very slippery.

Section 3: Composition and Information on Ingredients

Chemicals Characterization : molar ratio K_2O : SiO₂ = 1 : 3.15 to 1 : 3.33 Composition:

Name	CAS No.	EC No.	% by Weight
Potassium silicate	1312-76-1	215-199-1	39.4 - 41.0%
Water	7732-18-5	231-791-2	59.0 – 60.6%

	Section 4: First Aid Measures
Eye Contact:	Check for and remove any contact lenses. In case of contact, immediately flush eyes with plenty of water for at least 15 minutes. Get medical attention if irritation occurs.
Skin Contact:	Remove contaminated clothing and shoes. Wash with soap and water. Cover the irritated skin with an emollient. Get medical attention if irritation develops.
Inhalation:	If inhaled, remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention.
Ingestion:	Do NOT induce vomiting unless directed to do so by medical personnel. Never give anything by mouth to an unconscious person. If large quantities of this material are swallowed, call a physician immediately. Loosen tight clothing such as a collar, tie, belt or waistband.
Serious Ingestion:	Not available

	Section 5: Fire and Explosion Data
Flammable limits:	This material is noncombustible.
Extinguishing Media:	This material is compatible with all extinguishing media
Hazards to fire-fighters:	See Section 2 for information on hazards when this material is present in the area of a fire.
Fire-fighting equipment:	The following protective equipment for fire fighters is recommended when this material is present in the area of a fire: chemical goggles, body-covering protective clothing, chemical resistant gloves, and rubber boots.
Personal protection:	Wear chemical goggles, body-covering protective clothing, chemical resistant gloves, and rubber boots or safety shoes. See section 8.
Environmental Hazards:	Sinks and mixes with water. High pH of this material is harmful to aquatic life.

Section 6: Accidental Release Measures		
Small Spill:	Mop up and neutralize liquid, then discharge to sewer in accordance with federal, state and local regulations or permits.	
Large Spill:	Keep unnecessary people away; isolate hazard area and deny entry. Do not touch or walk through spilled material. Stop leak if you can do so without risk. Prevent runoff from entering into storm sewers and ditches which lead to natural waterways. Isolate, dike and store discharged material, if possible. Use sand or earth to contain spilled material. If containment is impossible, neutralize contaminated area and flush with large quantities of water.	
Personal protection:	Wear chemical goggles, body-covering protective clothing, chemical resistant gloves, and rubber boots or safety shoes. See section 8.	
Environmental Hazards:	Sinks and mixes with water. High pH of this material is harmful to aquatic life, see Section 12. Only water will evaporate from a spill of this material.	

Section 7: Handling and Storage		
Handling:	Avoid contact with eyes, skin and clothing. Avoid breathing spray mist. Keep container closed. Promptly clean residue from closures with cloth dampened with water. Promptly clean up spills.	
Storage:	Keep containers closed. Store in clean steel or plastic containers. Storage temperature 0-95 $^{\circ}$ C. Loading temperature 20-95 $^{\circ}$ C.	

	Section 8: Exposure Controls/Personal Protection
Engineering Controls:	Use with adequate ventilation. Keep containers closed. Safety shower and eyewash fountain should be within direct access.
Personal Protection :	Wear chemical goggles, body-covering protective clothing and gloves.

Section 9: Physical and Chemical Properties		
Appearance:	Liquid	
Odor:	Odorless.	
Color:	Clear to hazy white.	
Specific Gravity at 20°C:	1.388 – 1.401	
pH :	Approximated 11.8	
Boiling Point:	Not available	
Flash Point:	The product is not flammable	
Critical Temperature:	Not available.	
Oxidizing Properties:	None oxidizer	
Photodegradation:	No photodegradation	
Biodegradation:	Not applicable (inorganic substances)	
Solubility:	Soluble in cold water, hot water.	

Section 10: Stability and Reactivity Data		
Stability:	This material is stable under all conditions of use and storage.	
Conditions of Instability:	None.	
Materials to avoid	Gels and generates heat when mixed with acid.	
Hazard decomposition product:	None.	

Section 11: Toxicological Information		
Acute Data:	When tested, potassium silicates MR 2.25 Density 1.32 100% solids basis, its single dose acute oral LD50 in rats is 5,700 mg/kg	
Skin Irritation Data:	The irritation response is inversely correlated with the molar ratio of the silicates; a lower molar ratio SiO2 : K2O leads to a higher irritation score and vice versa. This correlation is superimposed by the concentration effect: lower concentrations will exhibit lower irritancy as compared to higher concentrations of the same molar ratio. Whereas ratios of 2.0 - 3.0 mild irritating properties, ratios over 3.0 were not irritating.	
Eyes Irritation Data:	Varying molar ratios exhibited effects in enucleated rabbit eyes ranging from corrosive MR lower than 3.4 to slightly irritating , MR over 3.4 were not irritating	
Subchronic Data:	No information available for Potassium Silicate. But in Sodium Silicate study shows that no changes to rats' organ when feed for six month NOAEL (180d) at 159 mg/kg bw/d	

	Section 12: Ecological Information Ecotoxicity
Ecotoxicity:	The following data is reported for chemically similar potassium silicates on a 100% solids basis: LC50 (48 hour) for fish (Leuciscus idus) of 146 ppm at MR 3.9-4.1; EC50 (24 hour) for water fleas (Daphnia magna) of 146 mg/l at MR 3.9-4.1
Environmental Fate:	This material is not persistent in aquatic systems, but its high pH when undiluted or unneutralized is acutely harmful to aquatic life. Diluted material rapidly depolymerizes to yield dissolved silica in a form that is indistinguishable from natural dissolved silica. It does not contribute to BOD. This material does not bioaccumulate except in species that use silica as a structural material such as diatoms and siliceous sponges. Where abnormally low natural silica concentrations exist (less than 0.1 ppm), dissolved silica may be a limiting nutrient for diatoms and a few other aquatic algal species. However, the addition of excess dissolved silica over the limiting concentration will not stimulate the growth of diatom populations; their growth rate is independent of silica concentration once the limiting concentration is exceeded.
BOD5 and COD:	No COD or BOD impact on effluents.

Section 13: Disposal Considerations			
Classification:	Disposed material is not a Hazardous waste.		
Disposal Method:	Dispose in accordance with federal, state and local regulations and permits.		

	Section 14: Transport Information
DOT UN Status:	This material is not regulated hazardous material for transportation.
Identification:	Not applicable.
Special Provisions for Transport:	Not applicable.

Section 15: Other Regulatory Information					
HMIS (U.S.A.):	Health Hazard: 1 Fire Hazard: 0	Reactivity: 0 Personal Protection:			
National Fire Protection Association (U.S.A.):	Health: 1 Flammability: 0	Reactivity: 0 Specific hazard:			
Protective Equipment:	Chemical goggles, Body-c	covering protective clothing and gloves.			

Section 16: Other Information References:

Prepared by: Nichapa K.	Rev.04 1 August 20
OECD SIDS: Soluble Silicate, International Programme on Chemical Safety (IPCS): Che Information from Intergovernmental Organization (INCHEM), France 20-23 April 20	mical Safety 004
IUCLID Dataset Substance ID:1344-09-8, European Commission – European Chemicals 2000.	Bureau, 18 Feb

Soluble Silicate: Chemical, toxicological, ecological and legal aspects of production, transport, handling and application, European Chemical Industry Council, Belgium. Feb 2013

04 1 August 2017

ESIS: European chemical Substances Information System, http://esis.jrc.ec.europa.eu/, 29 May 2014

- International Chemical Safety Cards: Sodium Silicate (solution 25-50%), The National Institute for Occupational Safety and Health (NIOSH), http://www.cdc.gov/niosh/ipcsneng/neng1137.html, 20 June 2014
- Alphabetical index of substances and articles, United Nations Recommendations on the Transport of Dangerous Goods (UNRTDG) http://www.unece.org/fileadmin/DAM/trans/danger/publi/unrec/rev14 /English/05E_Index.pdf, 30 June 2014

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